

Chapter 1 Assignments

The following problems will be collected at the first exam.

Problems: 13,17,20,24,30,37,47,51,55,57,63,65,67,69,77,83,87,89,93,97,101,116,118,128

Exam Review Topics

terms

SI Units
Significant figures
Accuracy/precision
Physical/chemical changes

skills/operations

metric/metric conversions
U.S./metric conversions
Density problems
Temperature conversions

Chapter 2 Assignments

The following problems will be collected at the first exam.

Problems: 13,17,19,27,31,37,45,49,53,59,61,67,69,75,77,85,89,108,112,116

Exam Review Topics

terms

Electron/proton/neutron
Ionic/covalent
Atom/molecule
Isotopes
Atomic number
Mass number
"Pet names" for selected groups
Group/period
Metal/nonmetal/metalloid
Anion/cation

skills/operations

Determine total number of:
protons
neutrons
electrons
Predicting charge of an ion
Mole problems

Chapter 3 Assignments

The following problems will be collected at the first exam.

Problems: 4,17,23,29,33,37,41,43,51,59,63,69,75,85,89,91,95,99,103,110,114,124

Exam Review Topics

terms

Ionic/covalent bond
Polyatomic ion
Avogadro's number
Empirical formula
Percent composition

skills/operations

Mole concept (using N_a)
Calculating formula mass
Percent composition problems
Deriving empirical formulas
Naming ionic compounds
Naming covalent compounds

Chapter 4 Assignments

The following problems will be collected at the second exam.

Problems: 5,8,10,17,19,23,25,29,33,35,39,41,43,47,53,60,62,66,

Exam Review Topics

terms

Mole concept
Chemical Equation
Reactant/product

skills/operations

Balancing equations
Stoichiometric Calculations (incl. limiting reagent, %yield)
Combustion, Alkali Metals, and Halogen reactions

Chapter 5 Assignments

The following problems will be collected at the second exam.

Problems: 1,3,11,16,21,25,29,33,37,41,43,45,47,55,63,69,76,80,84

Exam Review Topics

terms

Solvent/solute
Solution concentration
Electrolyte..strong/weak
Ionic Equation
Net ionic equation
Salt
Oxidation number

skills/operations

Molarity/dilution problems
Writing Net Ionic equations
Precipitation reactions
Acid/base Neutralization reactions
Oxidation Reduction reactions
Solubility rules

Chapter 6 Assignments

The following problems will be collected at the second exam.

Problems: 6,13,17,25,29,31,35,39,45,51,55,59,63,67,73,79,87,91,100,106,118

Exam Review Topics

terms

Boyles Law
Charles Law
Gay-Lussac's Law
Avogadro's Law
STP
Dalton's Law of Part. Pressure
Mole fraction

skills/operations

Dynamic/static gas law problems
Molecular mass of a gas
Densities of gases
Mixtures of gases (Dalton's Law)
"Real" vs "Ideal" gases
Stoichiometry with gases
Diffusion/effusion of gases

Chapter 7 Assignments

The following problems will be collected at the second exam.

Problems: 13,18,26,35,41,47,51,57,61,65,67,73,75,79,81,85,89,100,106,112

Exam Review Topics

terms

Endo/exothermic
Enthalpy of reaction
Calorimetry
Heat capacity
Specific heat

skills/operations

Calorimetry problems
Hess' Law problems
Heats of Formation problems

Chapter 8 Assignments

The following problems will be collected at the third exam.

Problems: 4,11,14,16,37,39,43,51,59,61,69,71,80,83,100,104

Exam Review Topics

terms

Electron/proton/neutron
Uncertainty Principle
Wavelength/frequency
Quantum numbers
Line spectrum
de Broglie Relationship
Bohr Theory

skills/operations

Calculate frequency/wavelength
Calculate energy of a photon
Rydberg Equation
Shapes of orbitals

Chapter 9 Assignments

The following problems will be collected at the third exam.

Problems: 16,33,35,39,47,49,61,65,69,71,75,81,85,89,93,99,105

Exam Review Topics

terms

Hund's Rule
Pauli Exclusion Principle
Para/diamagnetic
Atomic radius
Ionization Energy
Electron Affinity
Electronegativity
Periodic Law

skills/operations

Electron configurations
Box diagrams
Quantum numbers
Trends in the Periodic Table

